

Name
Chemistry
Solubility Reactions #1

Molecular Equation: $AB + CD \rightarrow CB_{(aq)} + AD_{(s)}$	
Insoluble = precipitate Soluble = aqueous	Quick Check $Na^+ X$ $K^+ X$ $NH_4^+ X$ XNO_3^-
***Need at least 1 insoluble product or water	Always Soluble

- Determine potential products
- Indicate if the product will be soluble(s), insoluble (aq), or Liquid ($H_2O(l)$)
- If there are no "NEW" products indicate no reaction.

